TENNANTS DISTRIBUTION LIMITED

NORTHERN DIVISION
HAZELBOTTOM ROAD
CHEETHAM
MANCHESTER
M8 0GR
TEL 44(0)161 205 4454
FAX 44(0)161 203 4298
E-mail sales.manchester@tennantsdistribution.com



SOUTHERN DIVISION BOTANY WAY PURFLEET ESSEX RM19 1SN TEL 44 (0)1708 860075 FAX 44 (0)1708 860074

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PRODUCT DATA SHEET

(This booklet incorporates the Specification and M.S.D.S.)

PRODUCT	GRANU	LAR SALT	A
CAS NO.	7647-14-5		
TARIFF NO.	250100990		
U.N NO.	N/R		
EINECS NO.			
IMCO CLASS	N/R		
HAZARDS			
SPECIFICATION REFERENCE		SAGRA/1	DATE APR 01
REFERENCE NO.		SAGRA/2	DATE MAR 05
PREVIOUS EDITIO	N.	SAGRA/1	DATE APR 01

	PRODUCT SPECIFICATION	-		
Product Name	Salt (Granular, Type A)			
Alternative Name	Sodium Chloride			
Product Grade				
SALES SPECIFICATION				
Sodium Chloride (NaCl)	99.80% ds			
Humidity (H20)	0.09%			
Sulphate (S04)	1200 mg/kg			
Phosphate (P04)	<10 mg/kg			
Calcium (Ca)	400 mg/kg			
Magnesium (Mg)	300 mg/kg			
Potassium (K)	505 mg/kg			
Strontium (Sr)	26 mg/kg			
Boron (B)	15 mg/kg			
Aluminium (Al)	2 mg/kg			
Silicon (Si)	53 mg/kg			
Iron (Fe)	<1 mg/kg			
Chromium (Cr)	<0.2 mg/kg			
Copper (Cu)	0.3 mg/kg			
Mercury (Hg)	<1 mg/kg			
Iodine (I)	<2 mg/kg			
Insoluble Matter	0.01%			
	NOTES			

Exclusion of Liability

Information contained in this publication is accurate to the best of the knowledge and belief of Tennants.

Any information or advice obtained from Tennants otherwise than by means of this publication and whether relating to Tennants materials or other materials, is also given in good faith. However, it remains at all times the responsibility of the customer to ensure that Tennants materials are suitable for the particular purpose intended.

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Health and Safety

A Material Safety Data Sheet has been issued describing the health, safety and environmental properties of this product, identifying the potential hazards and giving advice on the handling precautions and emergency procedures. This must be consulted fully before handling, storage and use.

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SAFETY DATA SHEET IDENTIFICATION OF THE SUBSTANCE/PREPARATION AND COMPANY Product: Salt, Sodium Chloride (Granular, Type A) COMPANY: TENNANTS DISTRIBUTION LIMITED Hazelbottom Road Botany Way Cheetham Purfleet Manchester Essex M8 OGR **RM19 1SN** Tel No. 44(0)161 205 4454 Tel No. 44(0)1708 860075 Fax No. 44(0)161 203 4298 Tel No. 44(0)1708 860074 Emergency Tel No. 01865 407333 COMPOSITION/INFORMATION ON INGREDIENTS Chemical Formulae: NaCl Chemical Family Name: Sodium Chloride CAS No.: 7647-14-5 EINECS No.: 231-598-3 HAZARDS IDENTIFICATION 3. Inhalation Very high concentrations of salt dust may result in inflammation of the mucous membranes of the respiratory tract. Skin Contact Dry salt and concentrated solutions can cause withdrawal of fluid from skin and may, on prolonged contact, produce irritation. Eye Contact Salt and salt solutions are not toxic to the eye, but concentrations much above that of tears cause a stinging sensation Ingestion Acute and chronic toxic effects can result from the ingestion of excessive amounts of either salt or brine. Salt should not be used as an emetic to induce vomiting. High concentrations produce inflammatory reactions in the gastrointestinal tract and can cause vomiting, diarrhoea, convulsions and collapse. The ingestion of hypertonic solutions and cause fatal disturbances of body electrolyte and fluid balance, particularly in the young and elderly. Less than tablespoon of salt may severely poison an infant and sometimes prove fatal. FIRST AID MEASURES First Aid - Eyes Irrigate with eye wash solution or clean water. If symptoms develop obtain medical attention. First Aid - Skin Wash with plenty of water. First Aid - Inhalation Remove patient to fresh air. Keep warm and at rest. Give drinks if desired First Aid - Ingestion Vomiting will probably occur. Provided the patient is conscious give plenty of liquid to drink. Obtain immediate medical attention especially if vomiting has not occurred, FIRE FIGHTING MEASURES Flammability Non flammable Extinguishing Media Use agents suitable for type of surrounding fire, dry chemical, CO2, water spray Special Hazards Salt withstands temperatures up to its melting point and beyond without decomposing, but at very high temperatures (greater than approx. 800°C) a vapour may be emitted which is particularly irritation to the eyes Protective Equipment As applicable to the combustion products associated with the fire ACCIDENTAL RELEASE MEASURES Spillage Spillage should be swept up or may be safely water hosed to drain under normal circumstances Personal Protection Avoid prolonged contact with the skin and inhalation of dust concentrations, otherwise normal good handling and housekeeping practice is adequate. No special protective clothing is required. An eyewash bottle with clean water should be available HANDLING AND STORAGE Handling Salt dust is non flammable, but static electricity can be generated by pneumatic conveying, therefore pipes should be bonded and earthed, especially in environments where a spark could prove hazardous Storage Due to its hygroscopic nature, dried vacuum salt should be stored in a dry

atmosphere and away from concentrated acids. Absorbs moisture if the

	relative humidity is above 75%.		
8. EXPOSURE CONTROL	S/PERSONAL PROTECTION		
Occupational Exposure Limits	As total dust 10mg/m3 (8hr TWA)		
•	As respirable dust 5mg/m3 (8hr TWA)		
	None specified		
Dangerous Exposure	Static electricity can be generated by pneumatic conveying, therefore pipes		
Engineering Controls	should be bonded and earthed, especially in environments where a spark could		
	prove hazardous		
Personal Protection			
Respiratory Protection	If the process is such that salt dust is generated, a disposable face mask should		
	worn		
Hand Protection	Gloves to be worn if prolonged contact is anticipated. Dry salt and concentrated		
	solutions can cause withdrawal of fluid from the skin		
Eye Protection	Wear chemical safety goggles in situations where contact with the eyes may		
	occur		
Skin Protection	Skin should be washed to remove salt. Dry salt and concentrated solutions can		
	cause withdrawal of fluid from the skin		
Other Protective Measures	An eyewash and hand washing facilities should be readily available		
9. PHYSICAL AND CHEM	TICAL PROPERTIES		
Physical State	White granular crystals		
Boiling Point	1413°C		
Melting Point	801°C		
Density	1.1gm/ml		
Water Solubility	at 200°C 360 g/l		
10. STABILITY AND REA	CTIVITY		
Stability	Stable		
Hazardous Decomposition Products	Trace amount of hydrogen chloride gas may be evolved at temperatures in		
•	excess of 800°C. Contains no water on crystallisation. Does not react with		
	alkalis at ordinary temperatures		
Conditions to Avoid	Reacts with strong sulphuric acid or nitric acid to give hydrogen chloride gas		
Materials to Avoid	Under wet conditions can corrode many common metals, particularly iron,		
	aluminium and zinc. Stainless steel and model resist attack.		
11. TOXICOLOGICAL IN	FORMATION		
Eyes	Dusts may be irritating		
Skin	Irritation after prolonged contact		
Ingestion	Salt is an essential constituent of the diet. It provides important body		
	electrolytes and is the source of hydrochloric acid present in the gastric		
	juices. The blood stream contains nearly 1% sodium chloride. In normal		
	industrial use salt is not hazardous.		
	LD50 3000 mg/kg oral, rat		
Inhalation	Dusts may be irritating		
Carcinogenicity	Not considered to be a carcinogen		
Mutagenicity	Not considered to be a mutagen		
Reproductive Effects	None identified		
12. ECOLOGICAL INFOR	MATION		
Environmental Effect	A maximum value of 412 mg/l ensures the protection of all aquatic life.		
	Source: Water Research Centre - September 1990		
	96 hour LC (Fish) 6750 mg/l		
	48 hour EC (Daphnia) 2024 mg/l		
	72 hour IC (Algae) 3014 mg/l		
	Daphnia Sub acute 1062 mg/l		
	Fish Sub acute 433 mg/l		
	BOD 5 Day 0 mg/e		
	COD 0 mg/e		
	Earthworm Toxicity 1000 hg/cm2		
3. DISPOSAL CONSIDERA			
Disposal should be in accordance with lo			
14. TRANSPORT INFORM			
Not classified as Hazardous for Transpor	t		

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15. REGULATORY INFORMATION

This product not classified as dangerous for supply or conveyance

16. OTHER INFORMATION

The Specification has been amended. The Safety Data Sheets remain the same

Revision Date: 30/03/05